

# **WTERT Research Update**

## **Gasification of Biomass**

## **Solid Carbon Conversion**

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*WTERT 2008 Bi-Annual Meeting  
Columbia University  
New York, NY 10027*

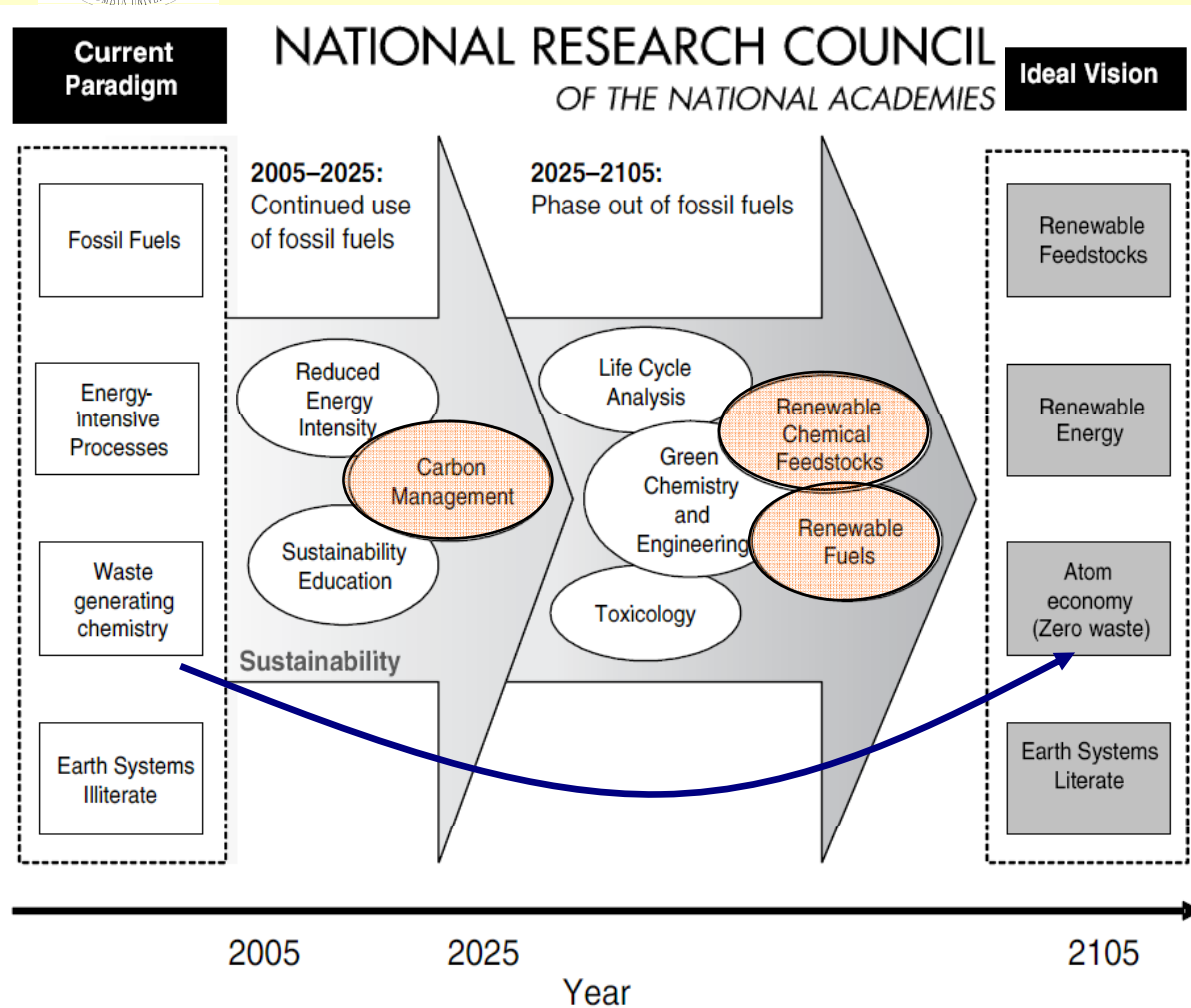
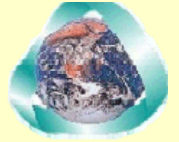
*October 16<sup>th</sup> and 17<sup>th</sup>*

WTERT 08, New York, NY



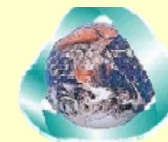
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# The Future



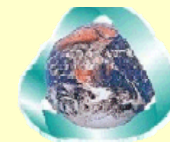
- Fossil Fuels will play smaller role
- Renewable / indigenous fuels will become more prominent
- Carbon management is here to stay

FIGURE ES-1 The Grand Challenges (ovals) for Sustainability (large arrows) that address the transition from current thinking to the ideal vision for the chemical industry over the next 100 years. See text for a more detailed description of figure and the Grand Challenges.



# The Need

- Carbon Neutral Energy Production
- Clean chemical (H<sub>2</sub>) Production
- Zero emissions → more than CO<sub>2</sub> sequestration
- Reduce the dependence on single feedstock
  - Indigenous source of fuel, distributed sources
- Power and chemicals produced from waste/biomass must be economically attractive as compared to current sources.



# Investment (wall Street)

- Socially responsible investing
- Emissions control (CO<sub>2</sub>)
- Eco-friendly business practices
- World Economic Forum
  - Businesses need to reduce the negative impact on the environment

*“Many companies are now discovering that there may be a strong correlation between ‘doing good’ and ‘doing well’”*

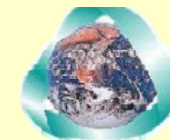
*“The next turn of the market wheel may reward companies with strong free cash flow as well as innovative companies thinking green”*

Alger Market Commentary March 31, 2007

WTERT 08, New York, NY

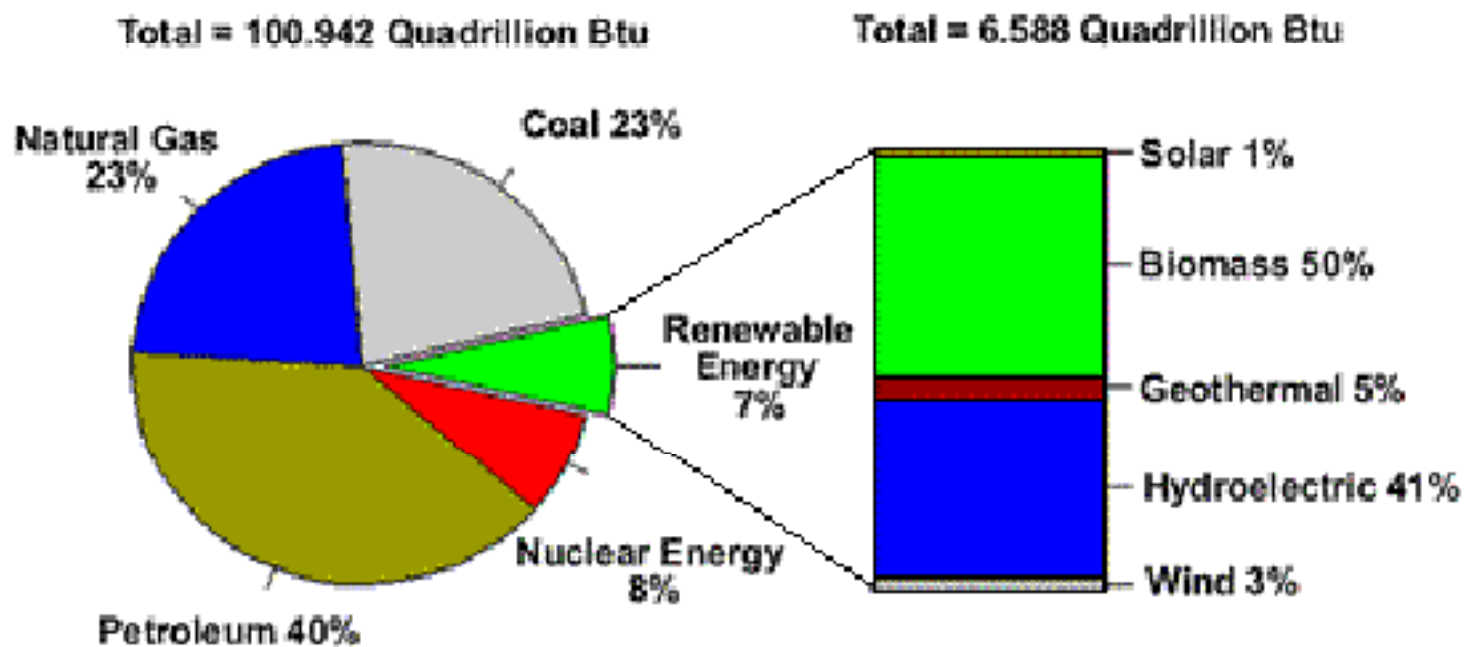


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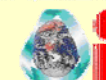
# Bioenergy

## The Role of Renewable Energy Consumption in the Nation's Energy Supply, 2005



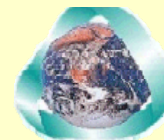
Source: IEA

## Renewable Portfolio Standards and State Mandates by State, 2007





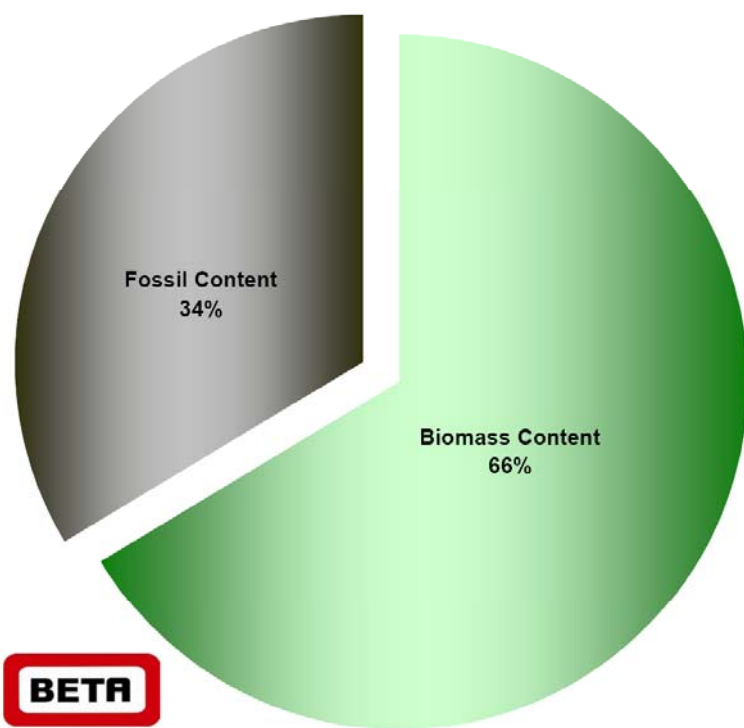
# Municipal Solid Waste (MSW)



- ~ 220 million tons of waste generated per year (U.S.)
- Landfills are filling up
- Disposal costs and energy costs are going up
- Greenhouse gas initiatives (RGGI, etc)

• **Waste as fuel emits 2/3 less CO<sub>2</sub> than fossil fuel**

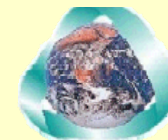
Proportions Biobased vs. Fossil Based  
indicated by <sup>14</sup>C content



**Potential to replace ~ 20% of oil imports per year**

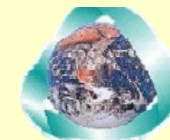
Burning is good (WTE); instead of oil → use waste

- Gasification gives options
  - choice of products – Heat, fuels, chemicals



# Chemicals From Waste

- Military MISER program
  - Trash/Biomass/Solid hydrocarbons to fuels
- American Chemical Society (ACS)
  - Letters to the editor – “chemicals from waste”
    - C&EN April 2006
- Discover Magazine –
  - “DATA” Section : The Ultimate Garbage Disposal
    - How to turn trash into clean energy – Geoplasma Unit
      - 160 MW by 2009, St. Lucie County, Florida
- EnerChem/City of Edmonton - 2008



# CO<sub>2</sub> & Waste / Biomass

## Enhanced gasification

Intermediates or fine chemicals for the chemical industry

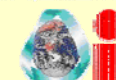
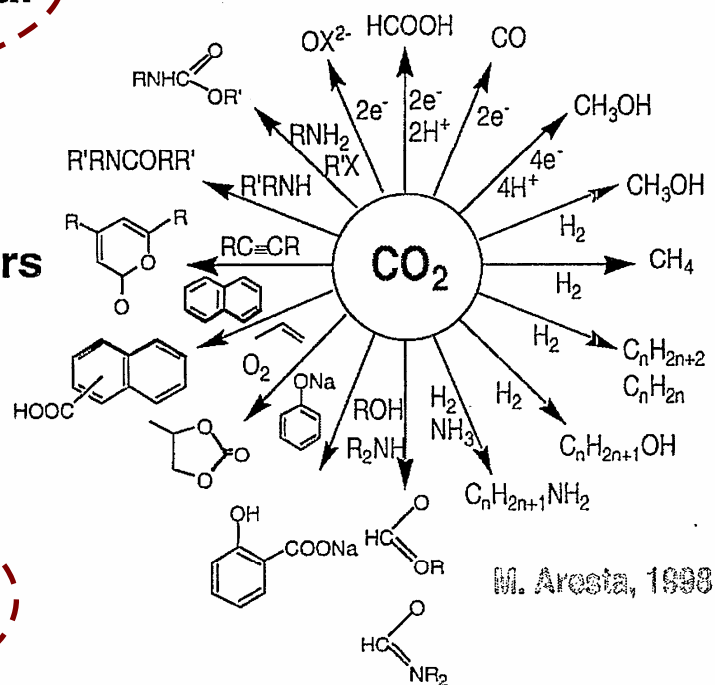
- C(O)O- : ~~Acids~~, esters, lactones
- O-C(O)O- : Carbonates
- NC(O)OR- : Carbamic esters
- NCO : Isocyanates
- N-C(O)-N : Ureas

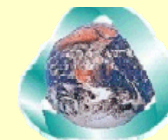
Use as a solvent

Energy rich products

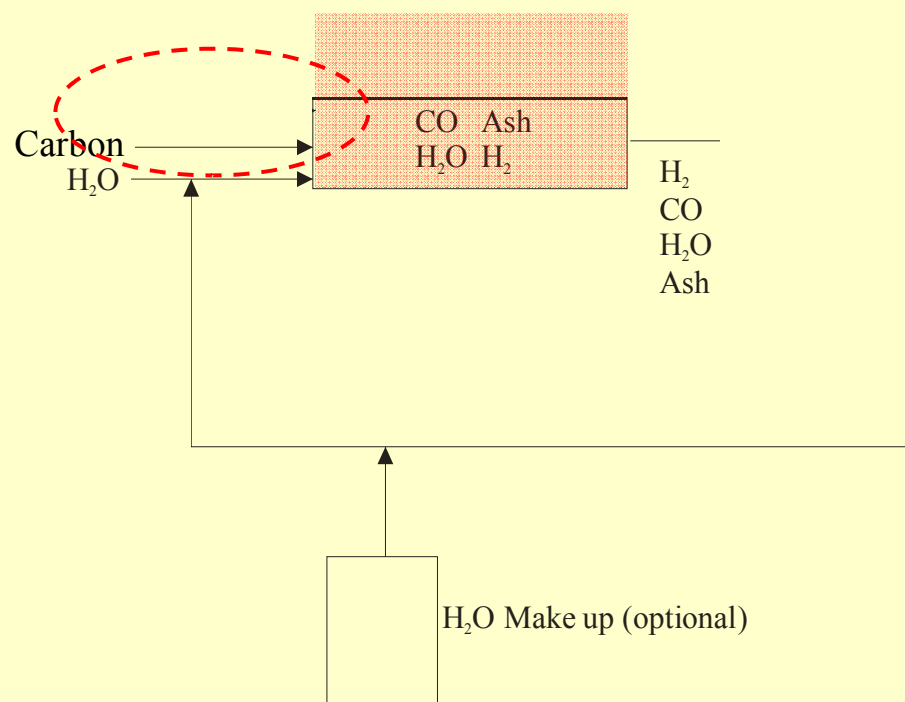
CO, CH<sub>3</sub>OH

Hydrocarbons and their derivatives





# Catalytically Controlled Reaction Gasifier (CRG Process)



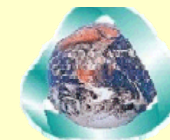
WTERT 08, New York, NY

Castaldi MJ, Dooher JP. *Int J Hydrogen Energy* (2007), 37, 4170-4179

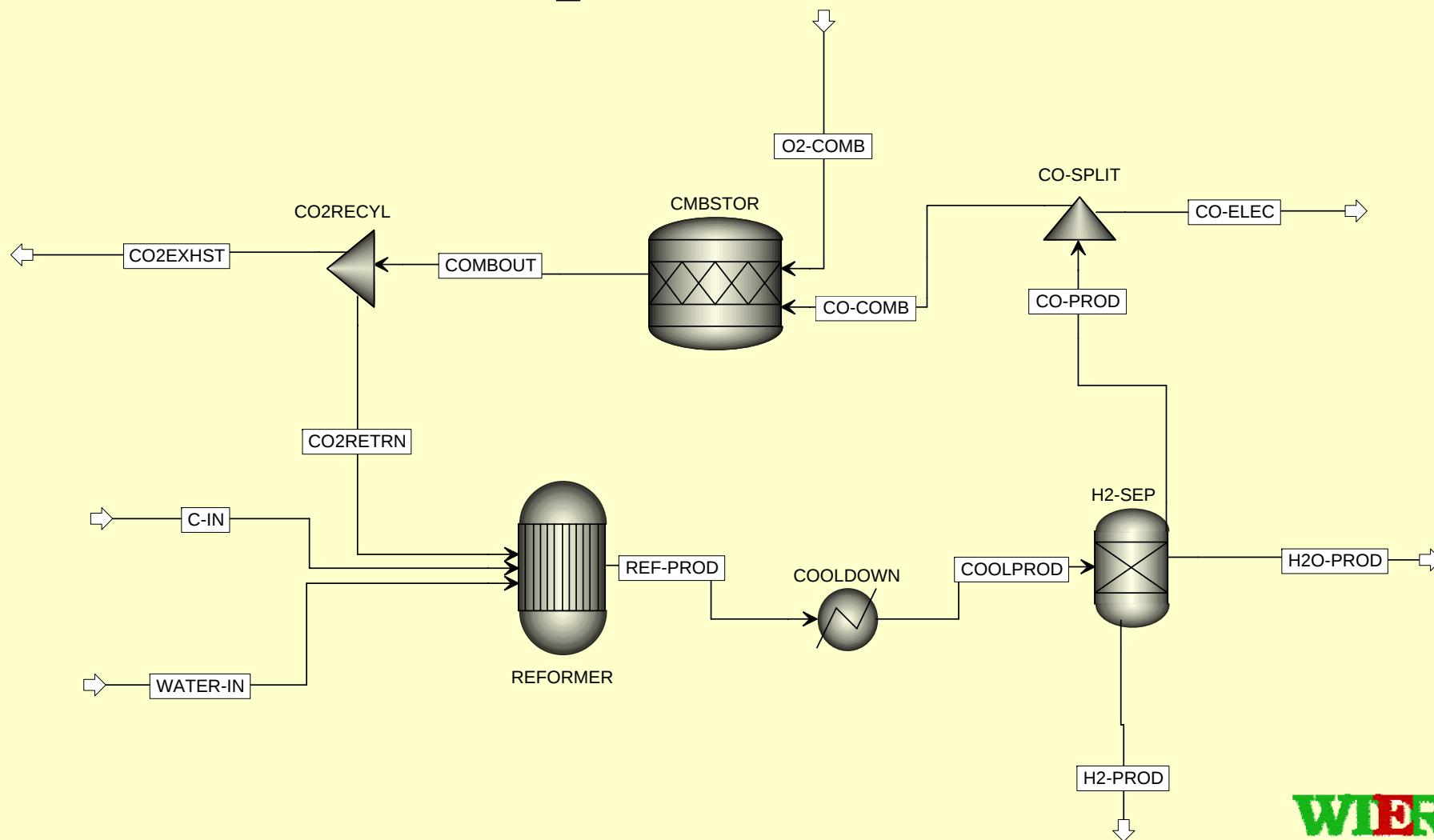
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# CRG Aspen™ Simulation





## Baseline simulation results of energy balance and hydrogen generation for CRG process

### Conventional Gasifier

$\text{H}_2 \rightarrow 0.7 \text{ kg/hr}$

$\text{CO}_2 \rightarrow \sim 15\%$

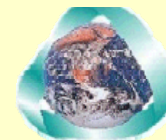
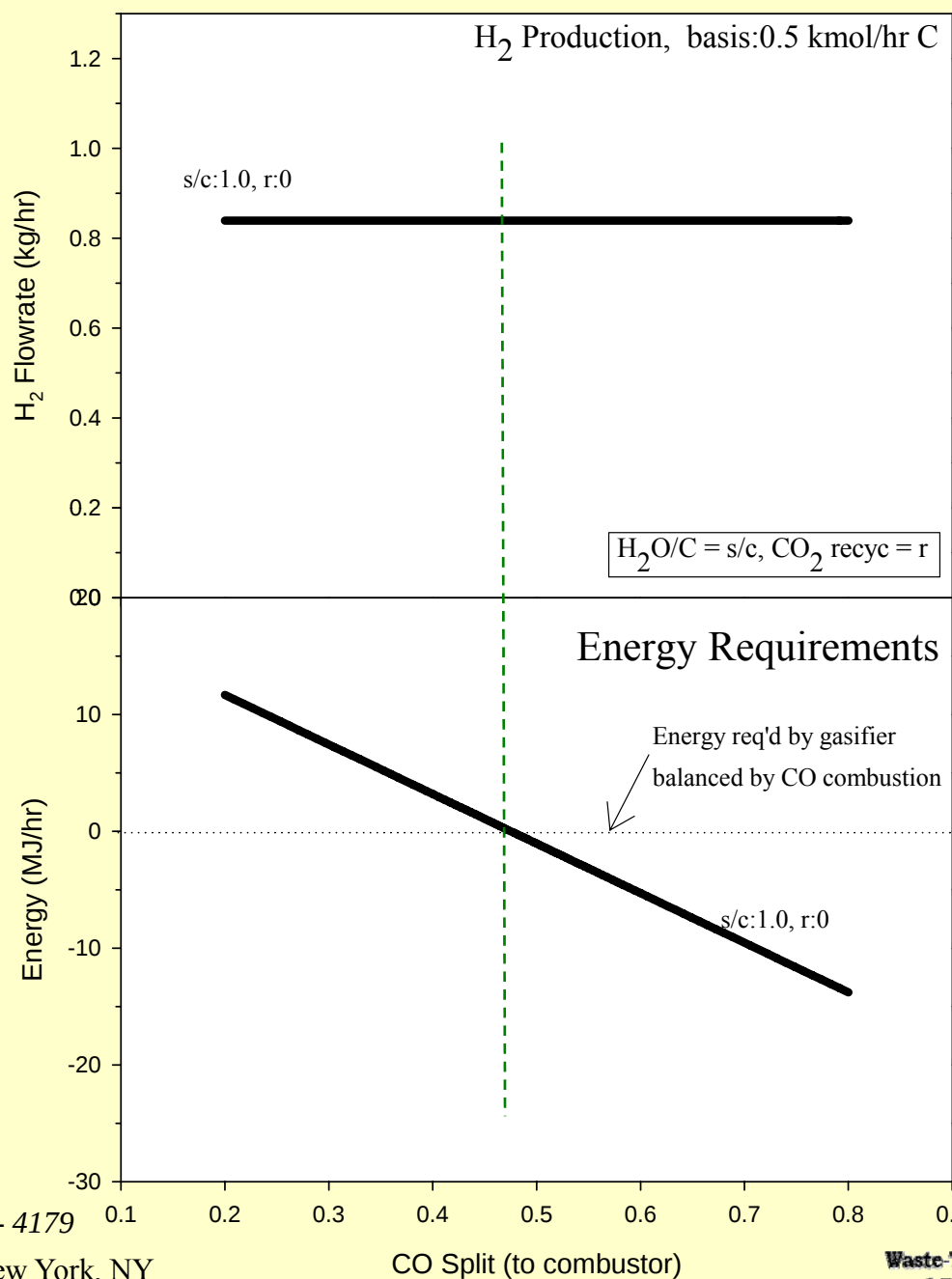
### CRG Process

$\text{H}_2 \rightarrow 0.85 \text{ kg/hr}$

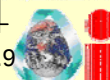
$\text{CO}_2 \rightarrow \sim 1.5\%$

$e^- \rightarrow \sim 13 \text{ kW/kmol C}$

$\text{CO Combustion} \rightarrow \sim 48\%$



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## Comparison of baseline and 25% CO<sub>2</sub> recycle to the reformer

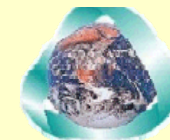
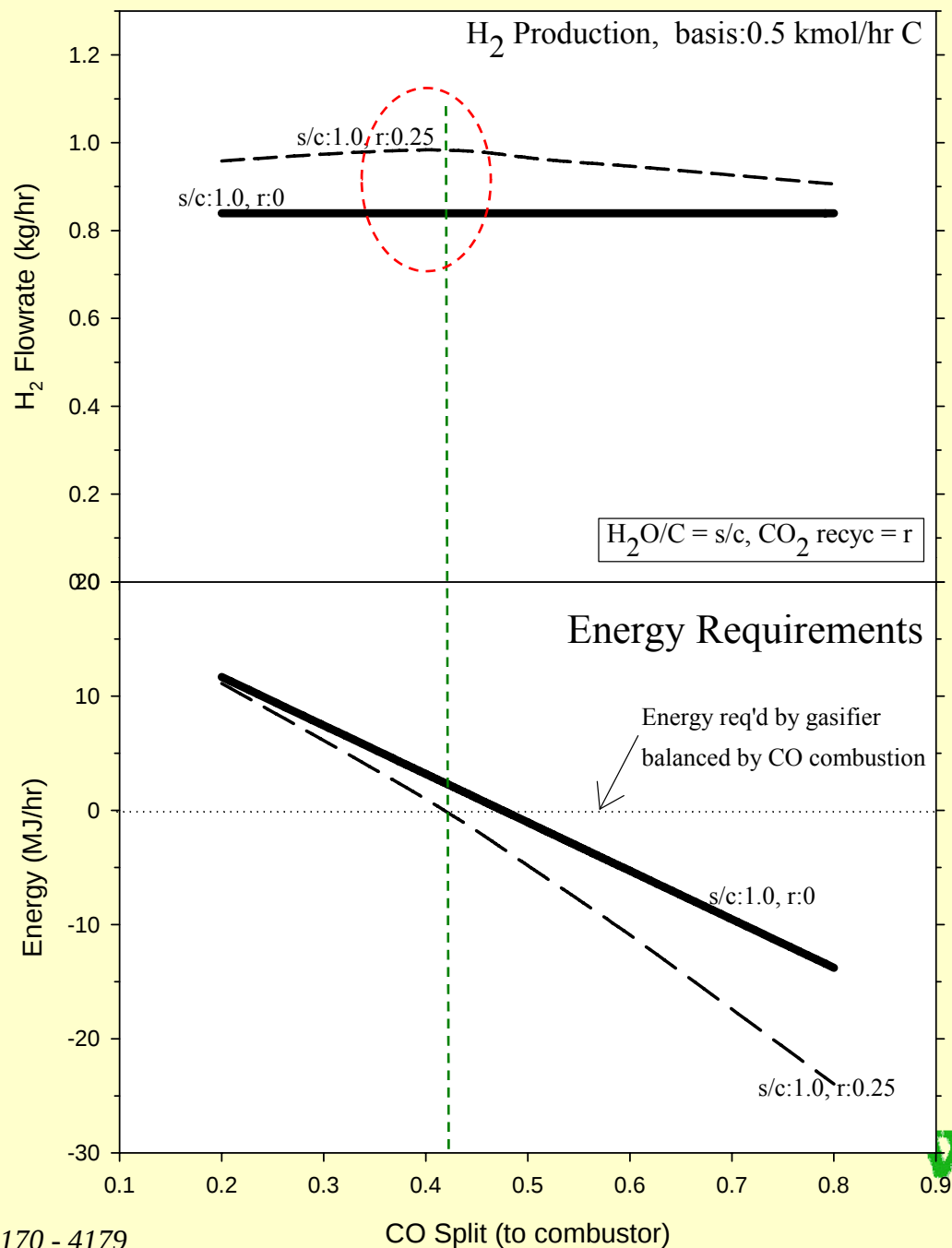
### CRG Process

H<sub>2</sub> → 0.95 kg/hr

CO<sub>2</sub> → ~ 1.5%

e<sup>-</sup> → ~13 kW/kmol C

CO Combustion → ~42%



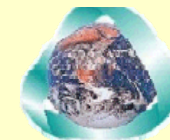
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# Experimental Feedstock Tested



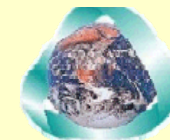
*Biomass  
Coal  
Waste*

| Biomass Fuel    | Classification | Mass % Residue | HHV (calories/gram) | Cellulose | Hemicellulose | Lignin |
|-----------------|----------------|----------------|---------------------|-----------|---------------|--------|
| Spruce          | Softwood       | 0.338          | 4719                | 48        | 21            | 28     |
| White Pine      | Softwood       | 0.316          | 4781                | 42        | 21            | 26     |
| Douglas Fir     | Softwood       | 0.086          | 4769                | 41        | 24            | 28     |
| Poplar          | Hardwood       | 0.253          | 4699                | 41        | 21            | 24     |
| Sugar Maple     | Hardwood       | 0.240          | 4725                | 39        | 30            | 23     |
| Red Oak         | Hardwood       | 0.110          | 4782                | 40        | 25            | 26     |
| Alfalfa         | Grass          | 5.890          | 4338 (ECN)          | 30        | 14            | 14     |
| Cordgrass       | Grass          | 4.360          | 4448                | 43        | 33            | 16     |
| Beachgrass      | Grass          | -              | 4329                | 45        | 31            | 12     |
| Wheat Straw     | Grass          | -              | 4402 (ECN)          | 38        | 29            | 14     |
| Maple Bark      | Bark           | 4.760          | 4316                | 18        | 33            | 19     |
| Blue Fir Needle | Needles        | -              | 4278 (ECN)          | 38        | 24            | 31     |
| Pine Needle     | Needles        | 4.130          | 4350 (IIT)          | 43        | 22            | 38     |
| Gr. Olive Pit   | Shell/Pit      | -              | 5165 (ECN)          | -         | -             | -      |
| Pecan Shell     | Shell/Pit      | -              | 4371 (SCERP)        | 38        | 23            | 29     |
| Almond Shell    | Shell/Pit      | -              | 4553 (ECN)          | -         | -             | -      |
| Walnut Shell    | Shell/Pit      | -              | 4823 (BEF)          | -         | -             | -      |
| Peanut Shell    | Shell/Pit      | -              | 4732 (BEF)          | 36        | 19            | 30     |
| Rice Hull       | Hull/Husk      | -              | 4575 (ECN)          | 6         | -             | 25     |

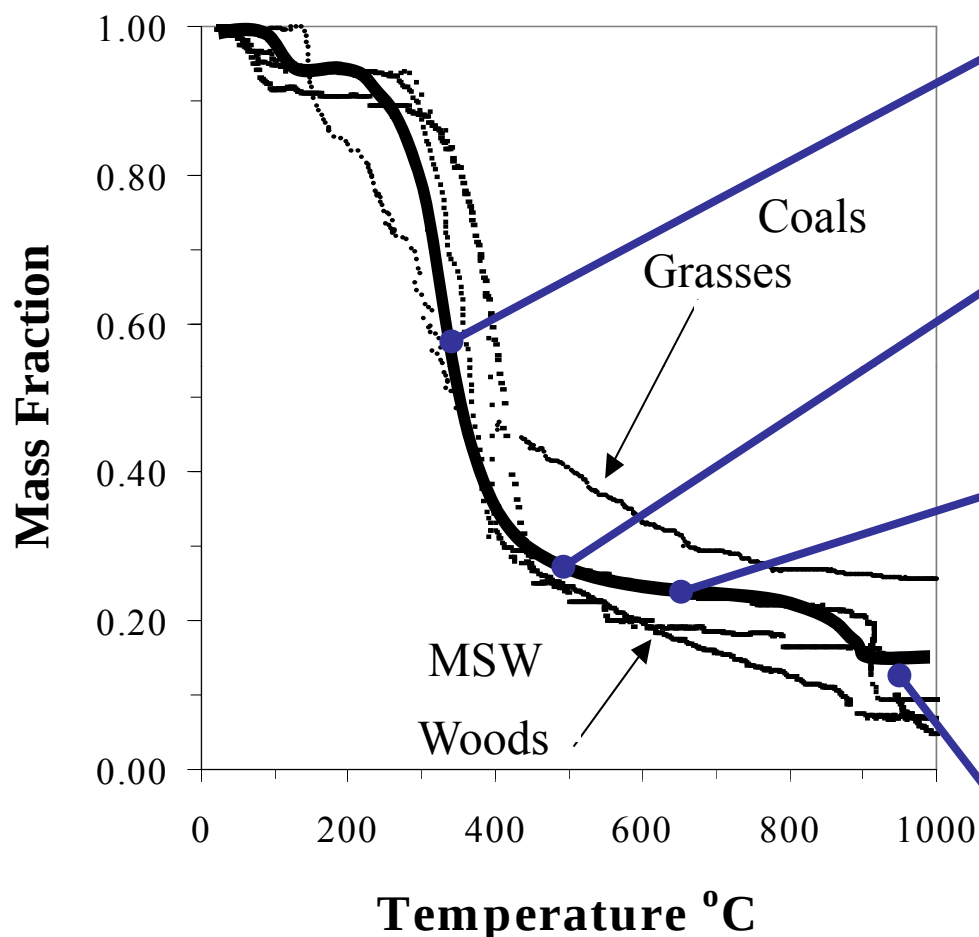
**Table 1.** Furnace combustion mass % residue, HHV from calorimetry and databases, and structural composition

| Coal Type | HHV (cal/gm) | Specific Gravity | %Ash | %Moisture | %C    | %H   | %O    | %N   | %Cl  | %S   |
|-----------|--------------|------------------|------|-----------|-------|------|-------|------|------|------|
| Wyoming   | 4690         | 1.26             | 5.43 | 29.59     | 49.22 | 3.41 | 11.02 | 0.77 | 0.01 | 0.31 |
| Montana   | 5176         | 1.10             | 4.12 | 25.40     | 54.14 | 3.80 | 11.50 | 0.71 | 0.00 | 0.34 |

**Table 2.** Specifications of the sub-bituminous coals taken from the quality spec sheets of the Cordero-Rojo mine in Wyoming and the Spring Creek mine in Montana



# Experimental Mass Decomposition Curve



## •250-400°C: Primary Reaction

- Bulk mass loss
- Devolatilization and pyrolysis

## •400-700°C: Secondary Reaction

- Ligno-cellulosic gasification

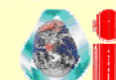
## •700-900°C: Tertiary Reaction

- Oxygen in feedstock & steam react with solid Carbon
- Recycled CO<sub>2</sub> begins char burnout

## •950 - 1000°C Mass burnout complete

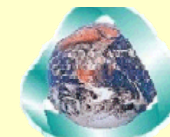
- Higher residual correspond to lower lignin content

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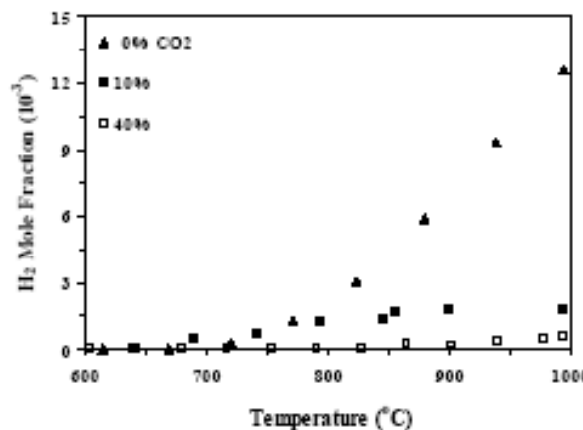


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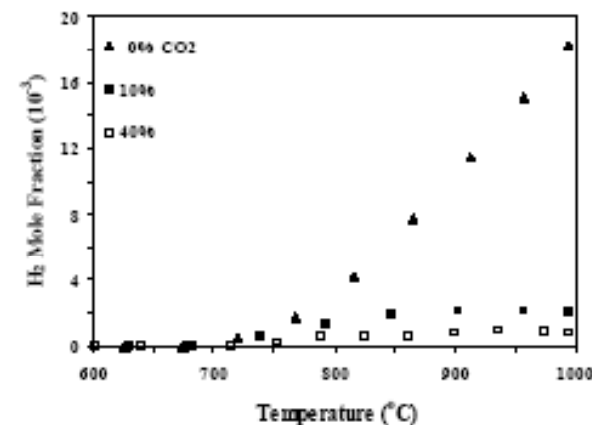
# CO<sub>2</sub> Impact on Coal



- **Decrease** in H<sub>2</sub> production with CO<sub>2</sub>

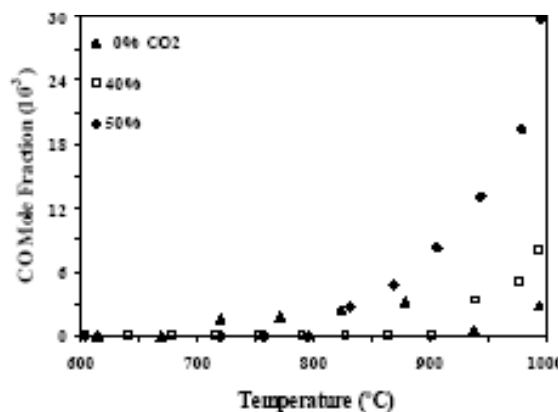


**Figure 21.** H<sub>2</sub> Evolution Depression from Montana coal using CO<sub>2</sub> (S/C = 10)

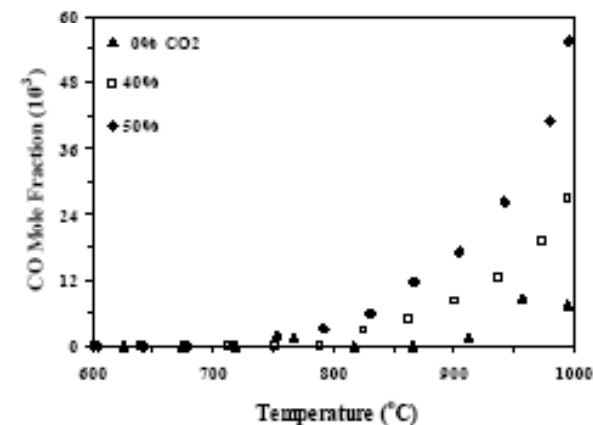


**Figure 22.** H<sub>2</sub> Evolution Depression from Wyoming coal using CO<sub>2</sub> (S/C = 9)

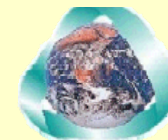
- **Increase** in CO production with CO<sub>2</sub>



**Figure 23.** CO Evolution Enhancement from Montana coal using CO<sub>2</sub> (S/C = 10)

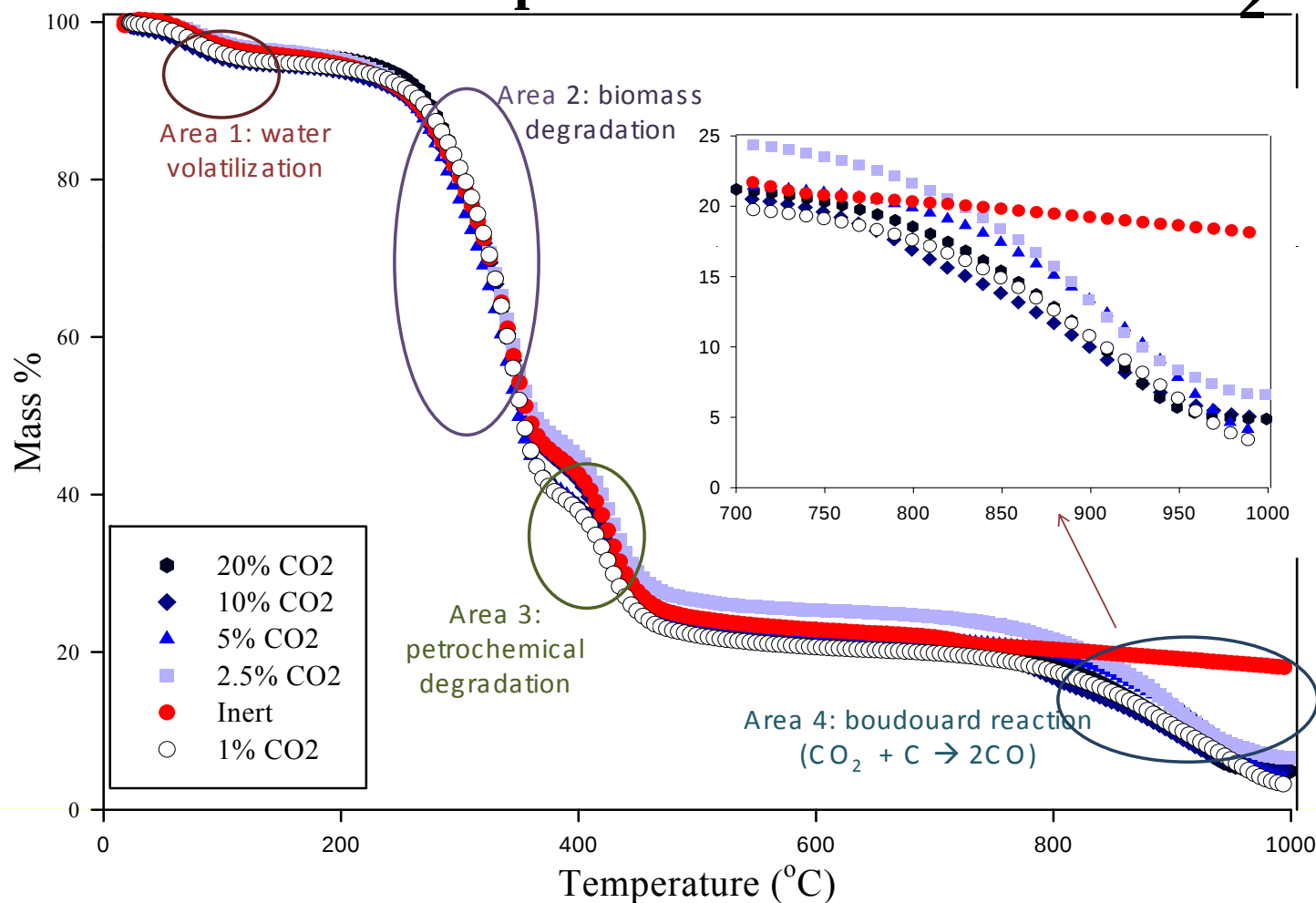


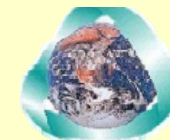
**Figure 24.** CO Evolution Enhancement from Wyoming coal using CO<sub>2</sub> (S/C = 9)



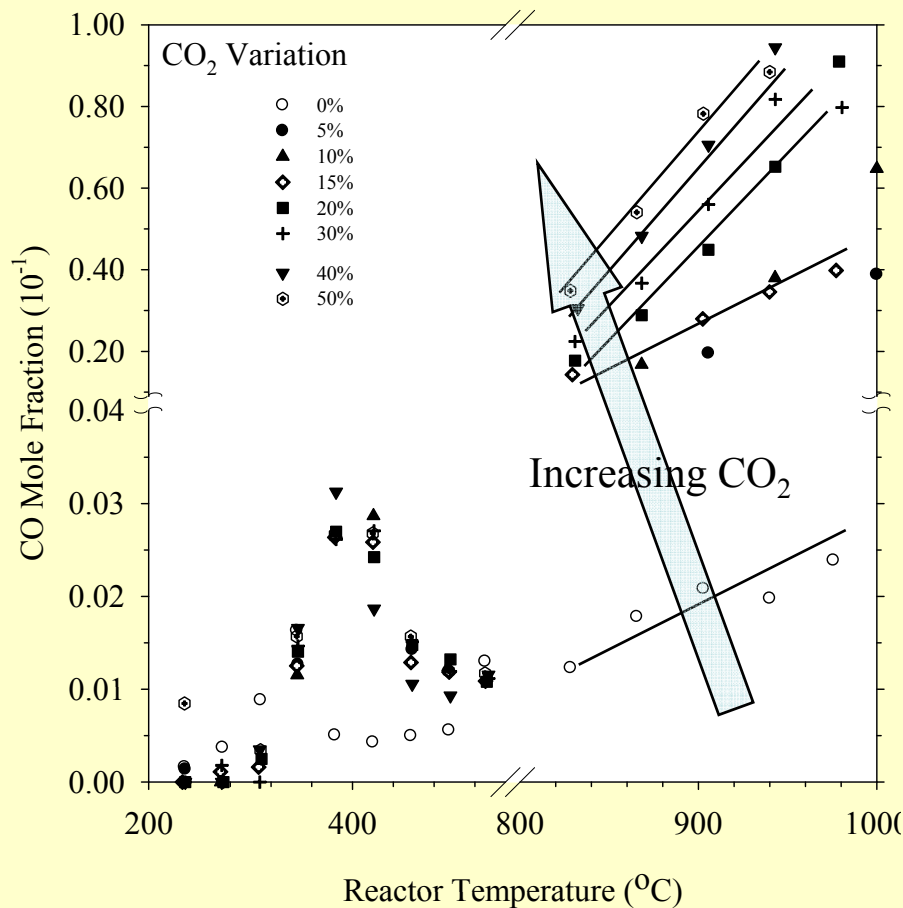
# MSW DATA

## Mass % vs Temp for Varius Amounts of CO<sub>2</sub>

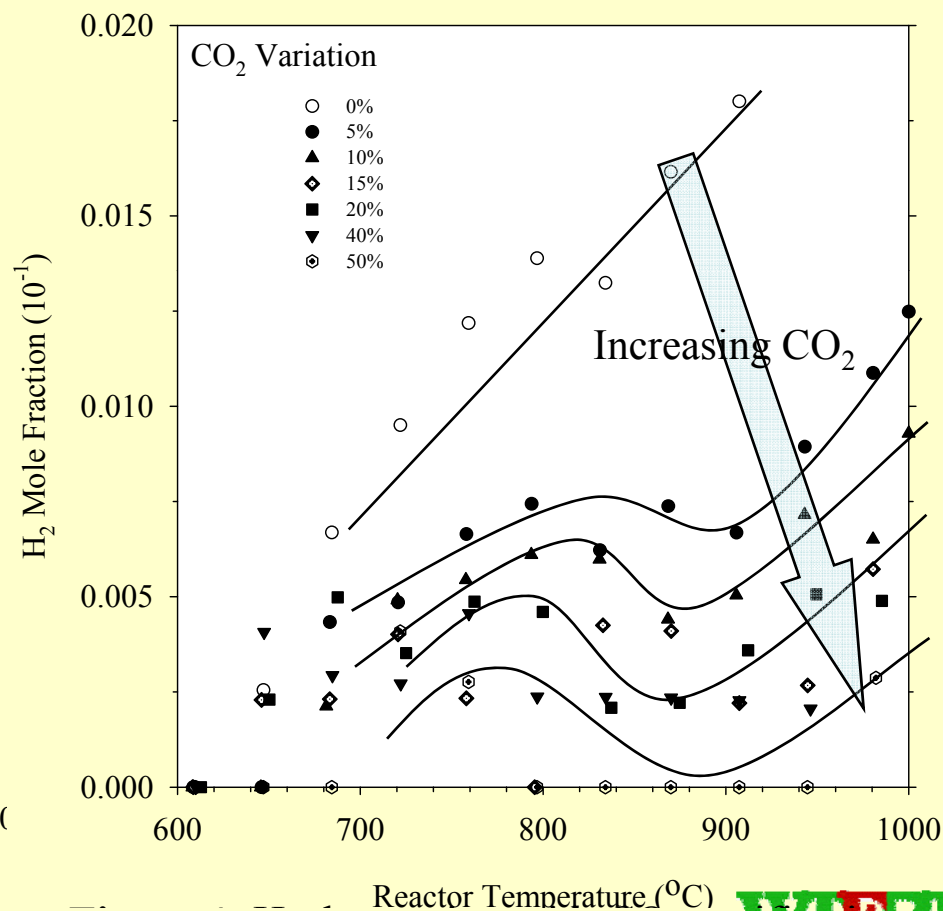




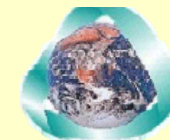
# CO<sub>2</sub> impact on gasification products



CO increases with CO<sub>2</sub>

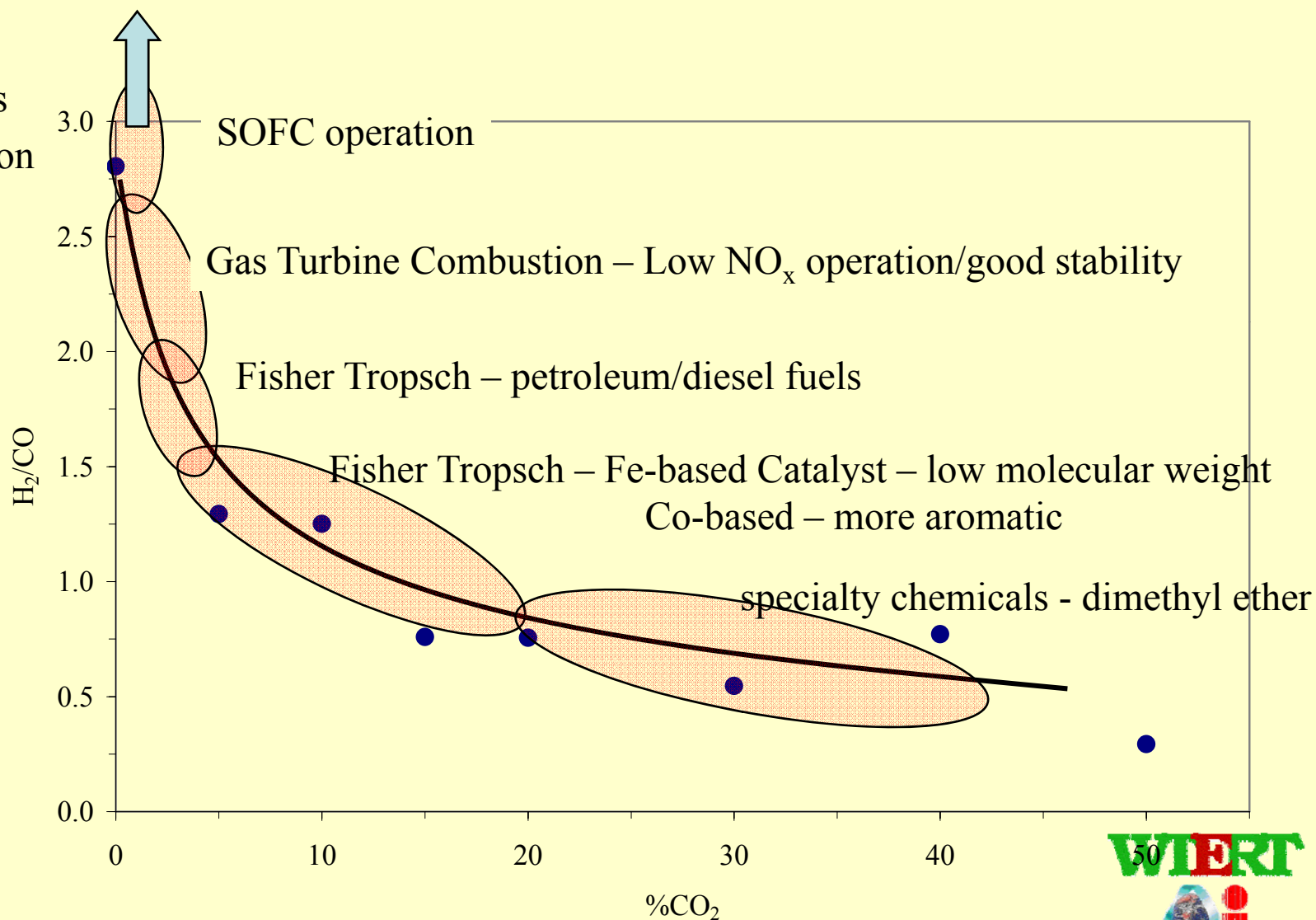


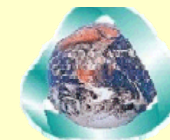
H<sub>2</sub> decreases with CO<sub>2</sub>



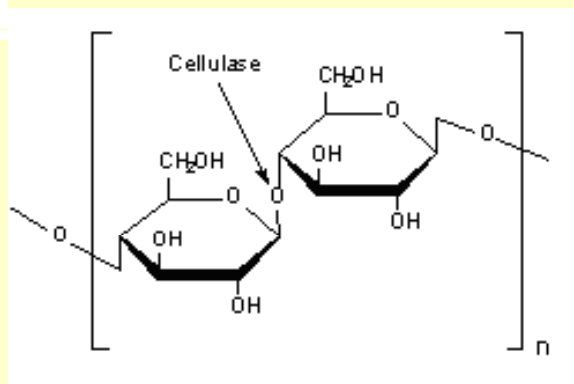
# H<sub>2</sub>/CO (Syngas) Tuning

- Fuels
- Chemicals
- Combustion
- Fuel cells

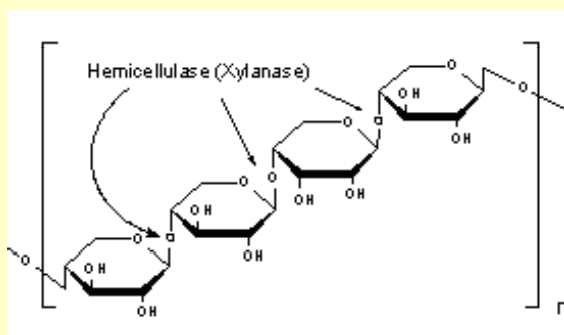




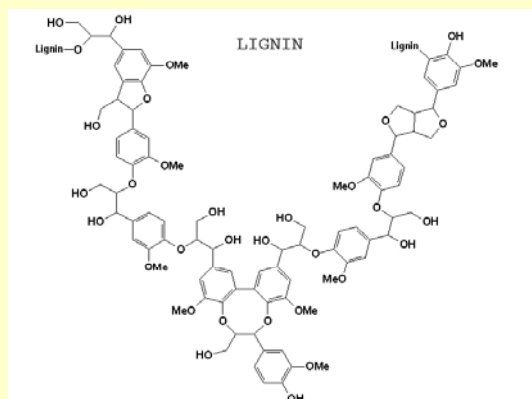
# Structural Components of Biomass



Cellulose- the main structural component of biomass, linear crystalline polymer of glucose held together by  $\beta$ -glycosidic linkages strengthened by H-bonding between chains

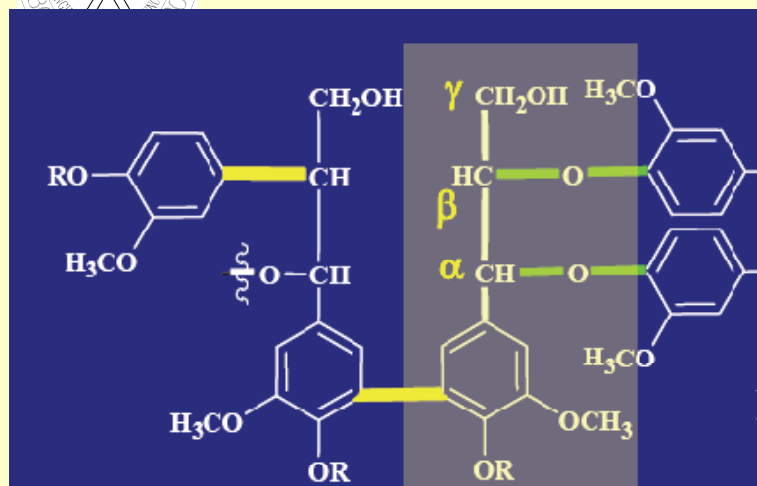
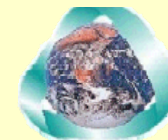


Hemicellulose- short highly branched, less rigid and more easily hydrolyzed, amorphous polymer of 5 and 6-carbon sugars

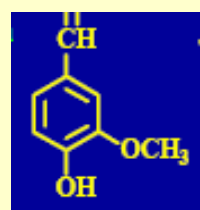
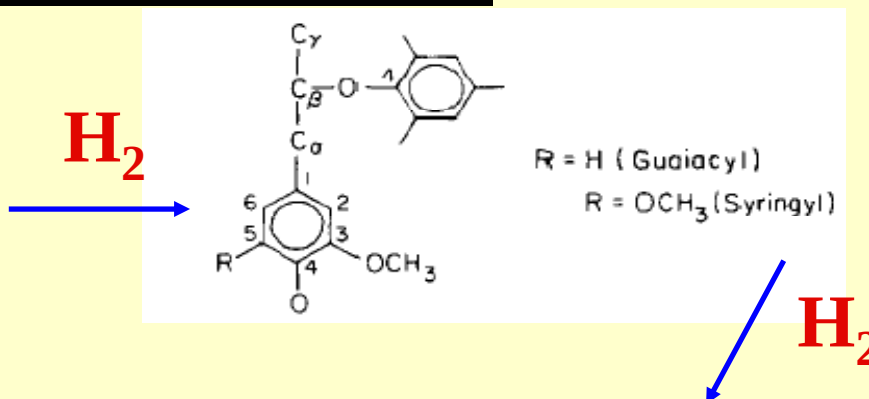


Lignin- the structural component of biomass that strengthens the cell wall and cements the cells together, high MW, highly cross-linked aromatic

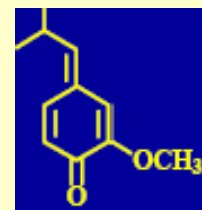
# Lignin Decomposition



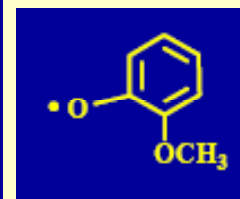
Lignin



Vinyl



Ionic



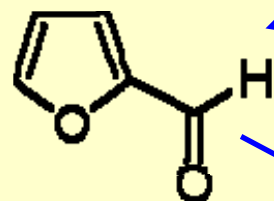
Phenoxy

$H_2$

$CH_4$   
Methane  
producing  
species

$CH_4$

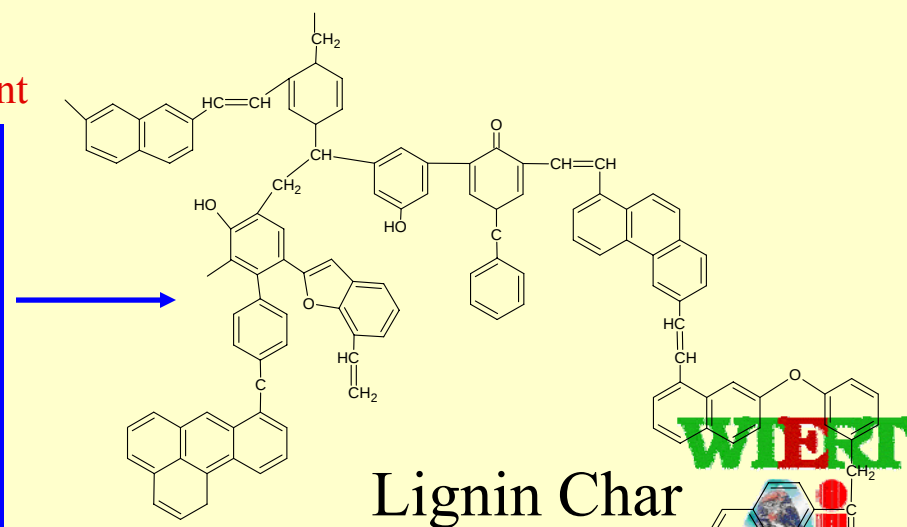
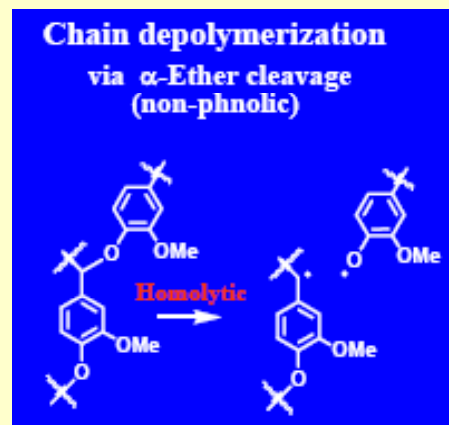
Continued Carbon enrichment



2-Furaldehyde  
Decarboxylation

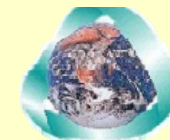
$CO$

Methylation and  
Dehydrogenation  
Of Lattice Structure

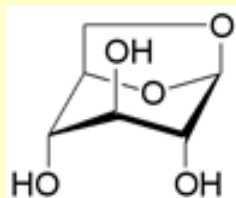


Lignin Char

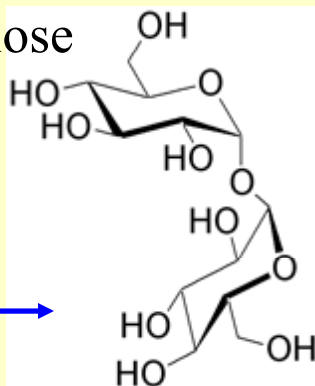
# Cellulose Decomposition



Levoglucosan

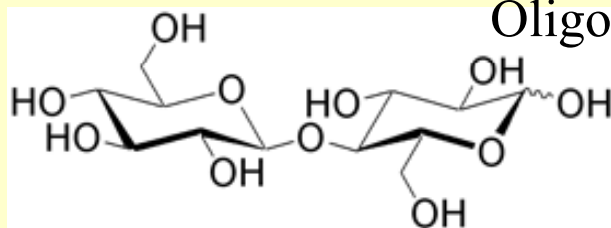


Trehalose



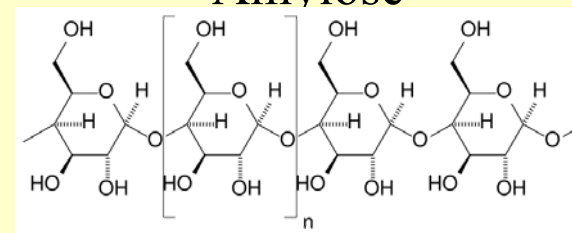
Dehydroxylation  
Polymerization

(Methanol soluble  
Oligosaccharides)

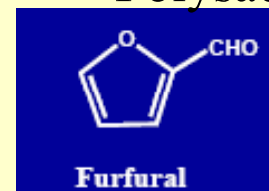


Cellobiose

Amylose



(Water soluble  
Polysaccharides and  
Furans and  
Carbonyls)

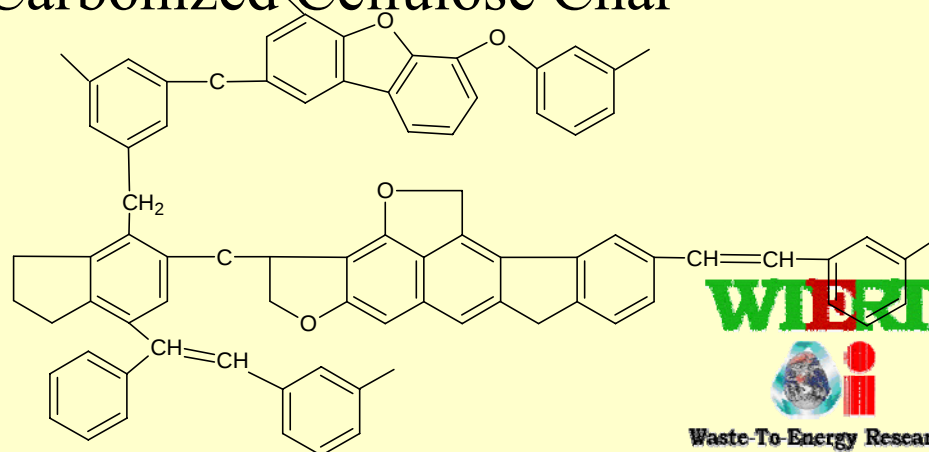


Furfural undergoes decarbonylation

Devolatilization  
No Char formation

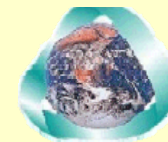
Low MW  
Volatiles  
 $\text{CO}, \text{CO}_2, \text{CH}_4, \text{H}_2$

Carbonized Cellulose Char



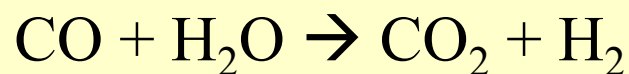


# Major Gasification Reactions

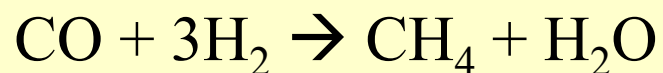
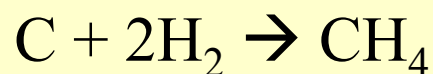


## Low Temperature

### Water Gas Shift



### Methanation



### Steam Gasification

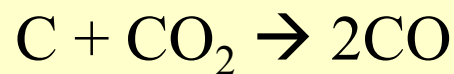


## High Temperature

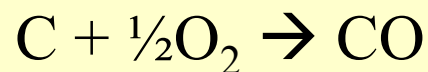
### Steam Gasification



### Boudouard

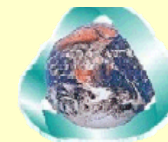


### Char Burnout: O (Biomass/Steam)

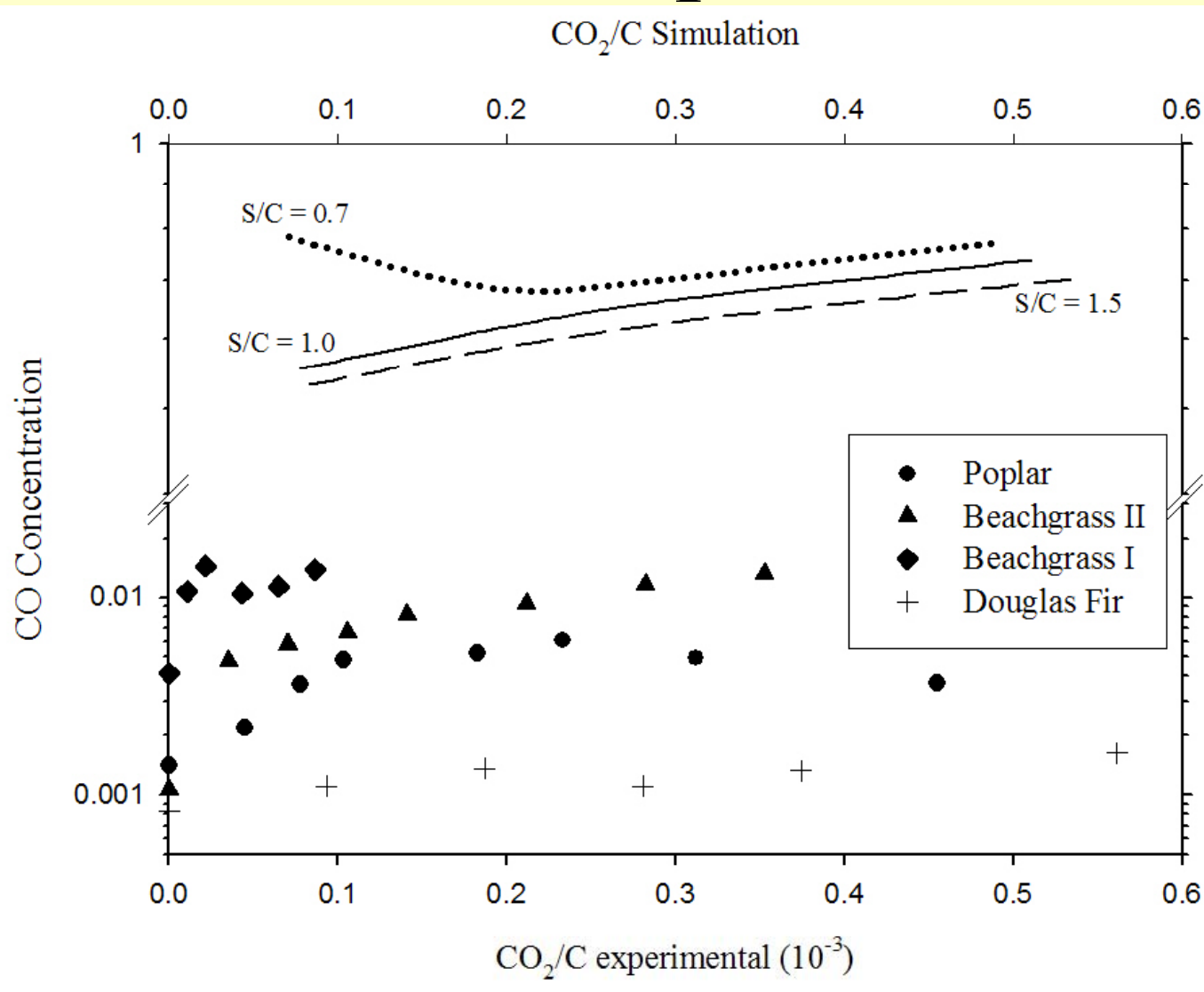


### Reverse Water Gas Shift



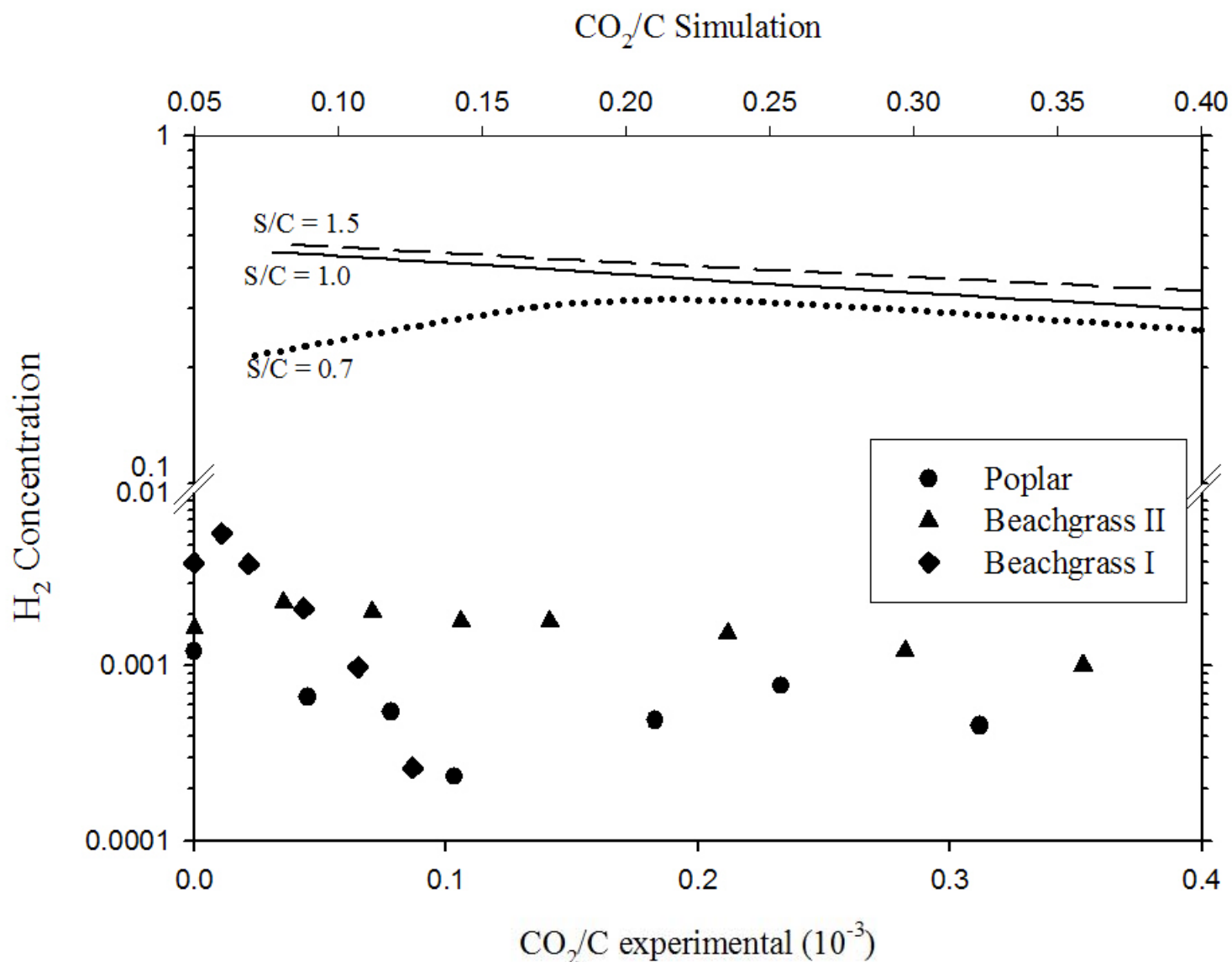
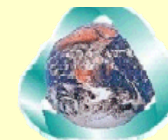


# CO Comparison



Butterman and Castaldi, *Indus. & Eng. Chem. Res.*, (2007) 47, 8875-8886

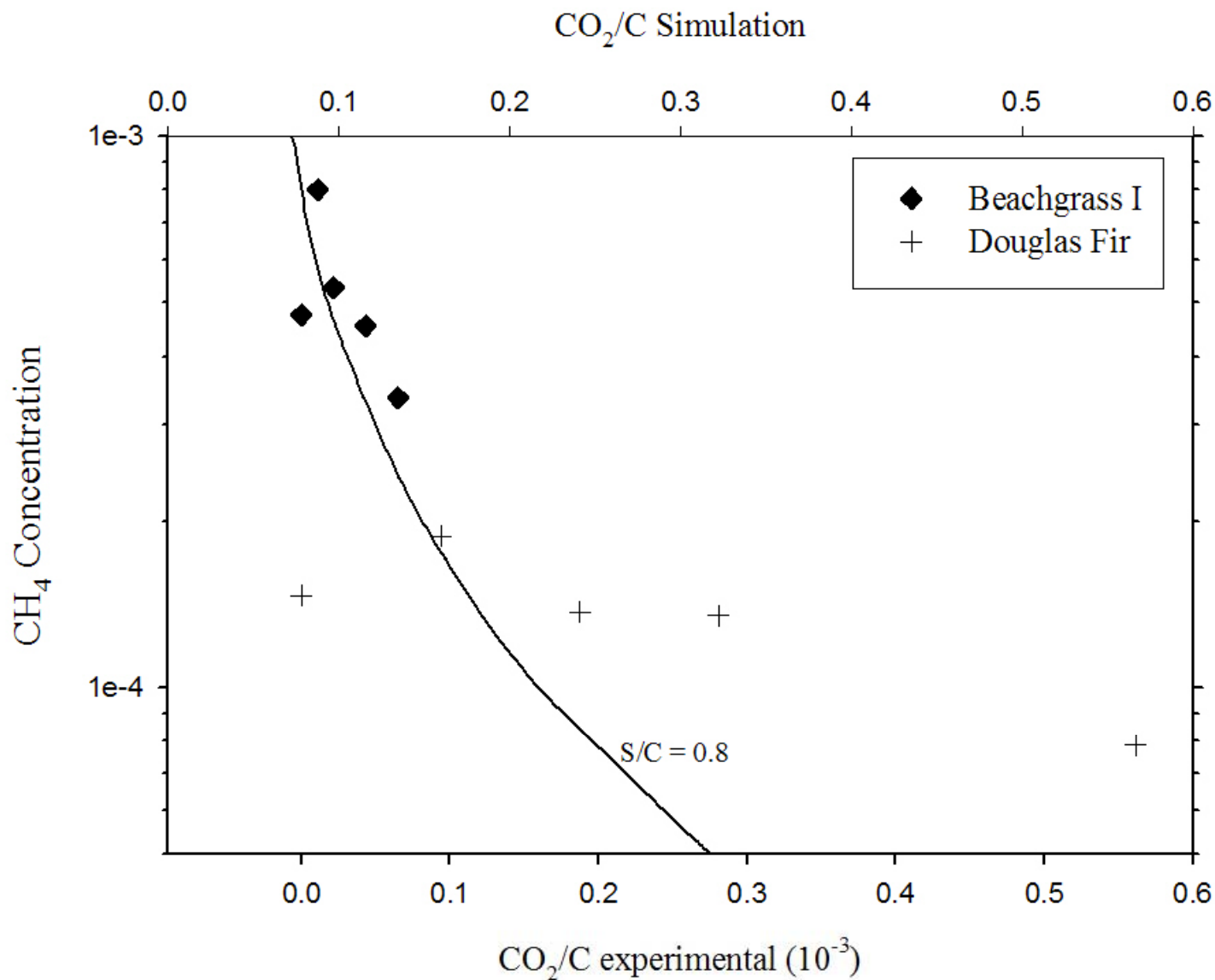
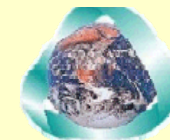
# H<sub>2</sub> Comparison

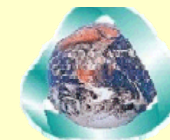


Butterman and Castaldi, *Indus. & Eng. Chem. Res.*, (2007) 47, 8875-8886



# CH<sub>4</sub> Comparison





## Value: Enhanced Char Burnout With CO<sub>2</sub>

- Identical time on stream, reaction temperature profile, total flow rate
- *Physical evidence of more efficient gasification with CO<sub>2</sub>*



Walnut Shells: 0% CO<sub>2</sub>



Walnut Shells: 30% CO<sub>2</sub>

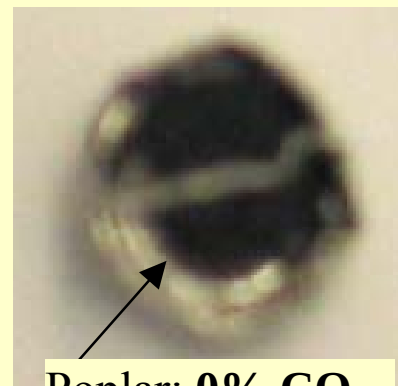


Douglas Fir: 0% CO<sub>2</sub>



Douglas Fir: 30% CO<sub>2</sub>

- Observed for all samples tested

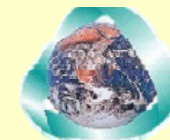


Poplar: 0% CO<sub>2</sub>



Poplar: 34% CO<sub>2</sub>

Butterman and Castaldi, Indus. & Eng. Chem. Res, (2007) 47, 8875-8886



# Char Pore Development- Enhanced Char Burnout With CO<sub>2</sub>



**Lignin**

**0% CO<sub>2</sub> -H<sub>2</sub>O/N<sub>2</sub>**

**1°C/min,**

**22-860°C**



**Lignin**

**100% CO<sub>2</sub>**

**1°C/min**

**22-860°C**



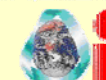
**Lignin**

**100% CO<sub>2</sub>**

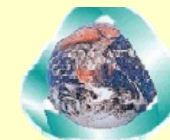
**1°C/min**

**22-930°C**

**WERT**

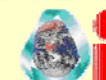
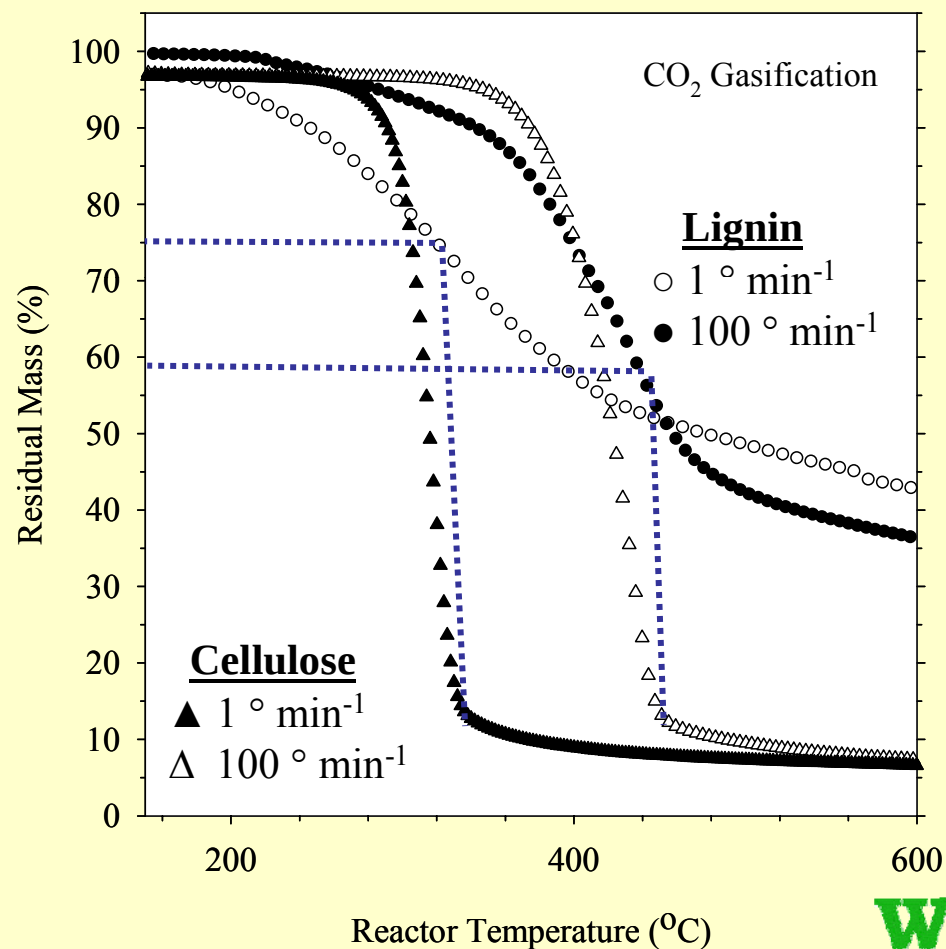


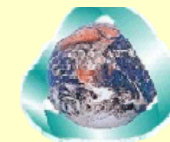
**Waste-To-Energy Research  
and Technology Council**



# Cellulose/Lignin Separation

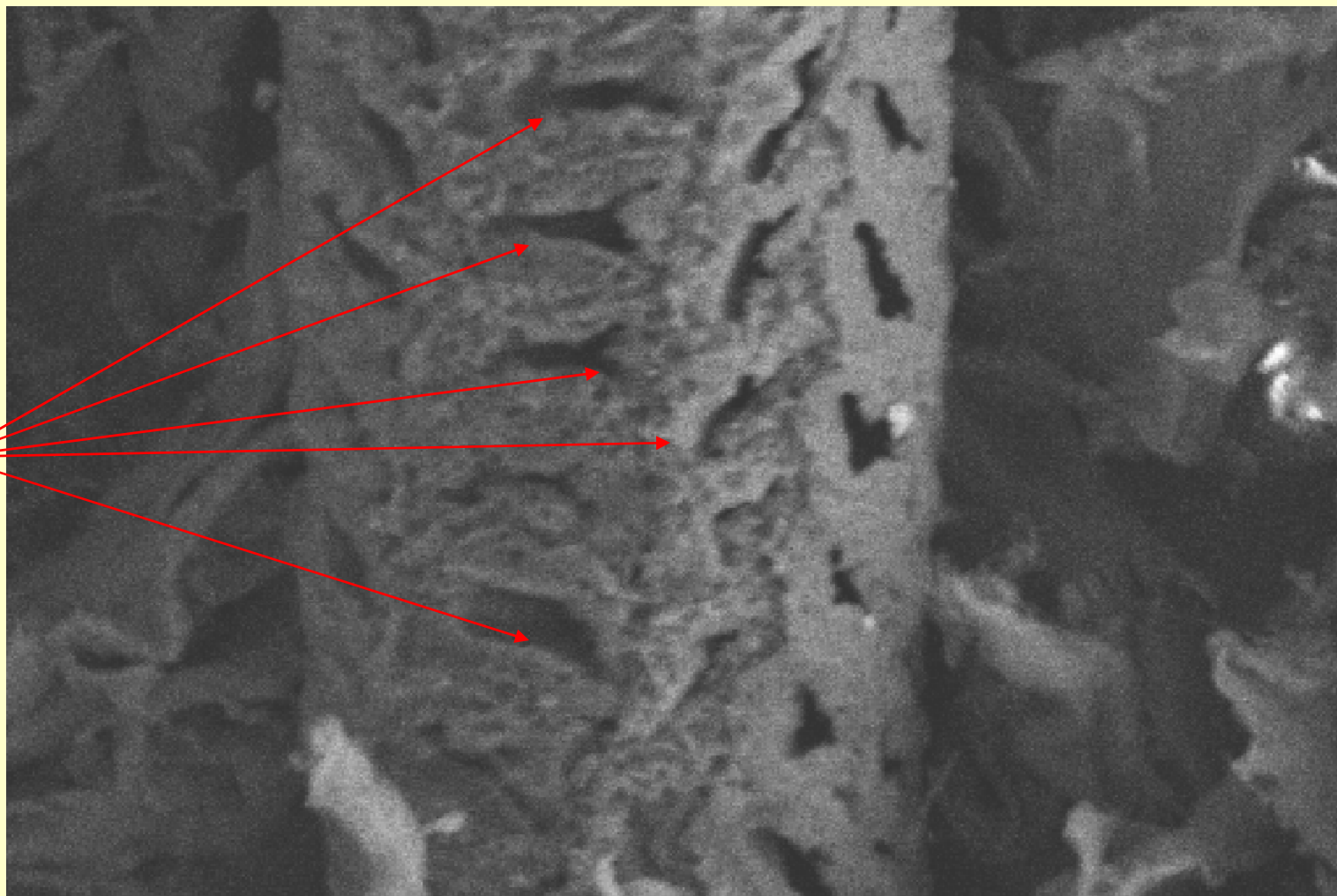
- Separation of lignin from cellulose, CO<sub>2</sub> processes lignin more efficiently than steam





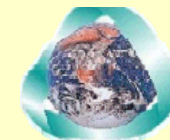
## SEM - Douglas Fir Char Fiber Enhanced micropore structure during steam gasification

**Increased porosity following thermal treatment**

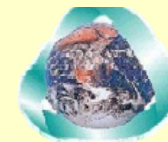




# Conclusions



- Data quantitatively matches Simulation
- CO<sub>2</sub> injection enhances CO production
  - T > 700°C resulting in improved char burnout
  - Less residual ash, especially carbon
- CO production near 400°C marks transition between the two thermal degradation regimes
  - 250-400°C – Local maxima of CO production
  - T > 700°C – Steady increase in CO
- H<sub>2</sub> production had two regions of abrupt production
  - 550-575°C for herbaceous
  - 675-725°C for woody feedstocks
- Coal and MSW follows trends of biomass
- CO<sub>2</sub> accesses the recalcitrant Char more effectively than steam



# Acknowledgements

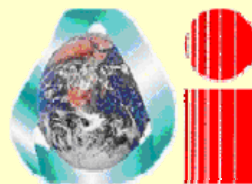
Dr. Heidi Buttermann – Post-Doc

Dr. Eilhann Kwon – PhD Student/Post-Doc

Kelly Westby – Undergraduate EEE student

Professor John Dooher – Adelphi University

# WERT



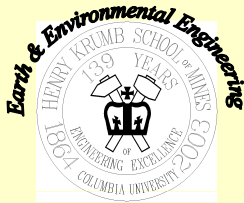
**Waste-To-Energy Research  
and Technology Council**

***Poster Session***

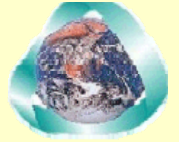
WERT 08, New York, NY



**Waste-To-Energy Research  
and Technology Council**



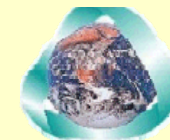
# Defining Sustainability



*Many equate sustainability with no emissions, or minimized raw material usage, with considering the impacts at the boundaries – avoided emissions or decreased energy use*

1. Sustainability is a path of continuous improvement, wherein the products and services required by society are delivered with progressively less negative impact upon the Earth – *IfS:AIChE*
2. Sustainability is an attempt to provide the best outcomes for the human and natural environments both now and into the indefinite future. - *Wikipedia*
3. Sustainable development as development that "**meets the needs of the present without compromising the ability of future generations to meet their own needs**". It relates to the continuity of economic, social, institutional and environmental aspects of human society, as well as the non-human environment - *Brundtland Commission*

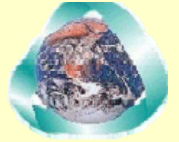
Sustainability is not a way to raise taxes – NYC mayor, \$8 to enter city by car during working hours



# Currently

***“The nature of environmental issues is changing from a regulatory to a resource focus” R. MacLean, Env. Protec. April 2003, P.12***

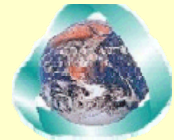
- **BP** → Re-staffing for environmental innovation
- **Walmart** → Supply chain vendors  
“20 chemicals of concern
- **Home Depot** → Eco-Label
- Carpet Recyclers profitable
- **GE** → wind power, clean water
- **NSF** → new directorate (EBW)
- **Venture Capital** → Niche energy/environment
- Education
  - Public Access to information
- Environment → resource utilization
  - Concentrators and Dilutors
- Climate change → economic issue



# Alternative Energy

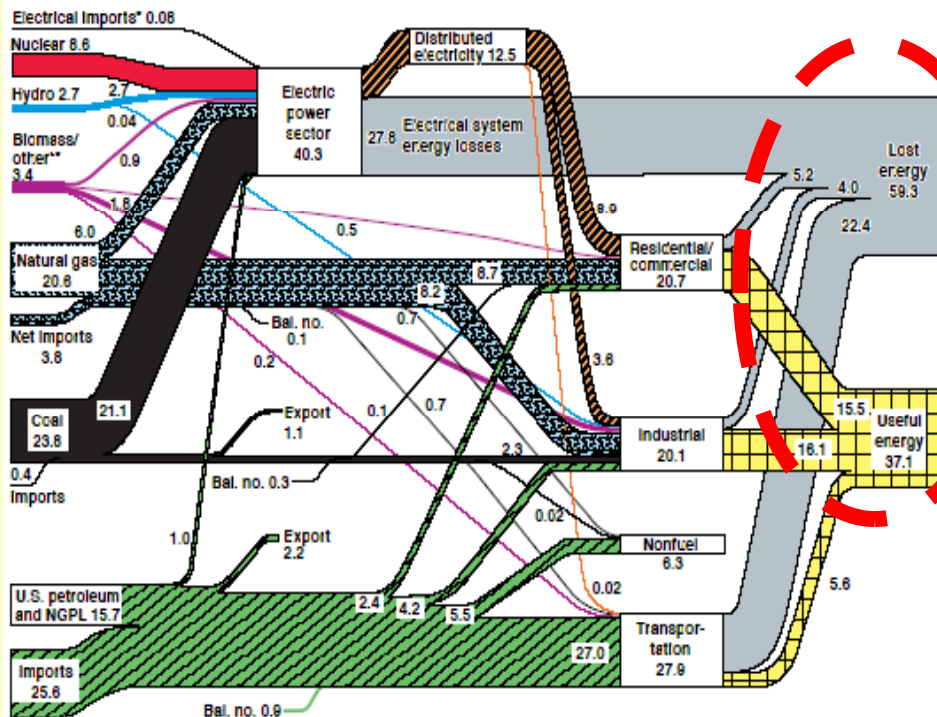
- The world is sensitive to energy supply (*new Paradigm*)
  - Security, Procurement → supply chain disruptions
- Two-fold increase in energy consumption (*demand*)
  - From 402 exajoules to 837-1041 over next 40 years.
- CO<sub>2</sub> atmospheric concentrations are rising (*environment*)
  - The need for separating carbon dioxide from the product gases and plant effluent in order to prevent it from entering the atmosphere.
  - Space constrained or preferred land use

# Scale of the Issue



## U.S. Energy Flow Trends – 2002

Net Primary Resource Consumption ~103 Exajoules

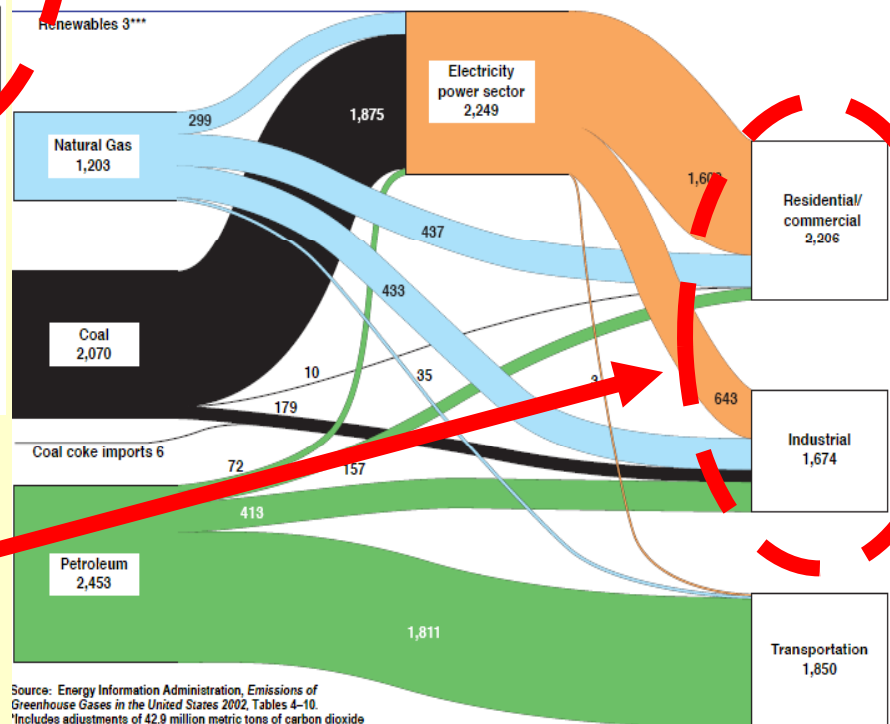


Source: Production and end-use data from Energy Information Administration, Annual Energy Review 2002.  
\*\*Net fossil-fuel electrical imports.  
\*\*\*Biomass/other includes wood, waste, alcohol, geothermal, solar, and wind.

June 2004  
Lawrence Livermore  
National Laboratory  
<http://eed.llnl.gov/flow>

- **Power / heat sectors are extremely inefficient**
  - ~60% of energy is wasted!
- **better power cycles and materials will help**
- **All fuels are needed!**

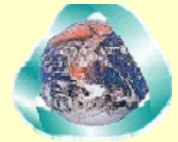
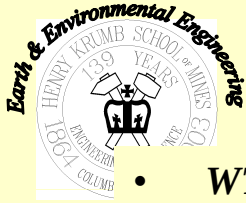
Fig. 7. U.S. 2002 Carbon Dioxide Emissions from Energy Consumption – 5,682\* Million Metric Tons of CO<sub>2</sub>\*\*



Source: Energy Information Administration, Emissions of Greenhouse Gases in the United States 2002, Tables 4-10.  
\*Includes adjustments of 42.9 million metric tons of carbon dioxide from U.S. territories, less 90.2 MtCO<sub>2</sub> from international and military bunker fuels.  
\*\*Previous versions of this chart showed emissions in metric tons of carbon, not of CO<sub>2</sub>.  
\*\*\*Municipal solid waste and geothermal energy.  
Note: Numbers may not equal sum of components because of independent rounding.

Lawrence Livermore National Laboratory, May 2004  
<http://eed.llnl.gov/flow>

- **Heat generation produces 2/3 of all CO<sub>2</sub> emissions**
- **Better thermal conversion systems needed**
- **Better Fuels needed!**



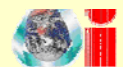
# Energy from Waste

- **WTE conserves fossil fuels by generating electricity. (Energy)**
  - 1 ton of waste combusted = 45 gallons of oil or 0.28 tons of coal
  - Most WTE facilities in U.S. process between 500 and 3,000 tons of waste per day
  - Electricity for 2.8 million homes
- **WTE facilities process 14% of the MSW in the United States. (Health)**
  - Trash-disposal needs about than 37 million people
- **WTE facilities meet some of the world's most stringent standards. (Environmental, local)**
  - Achieved compliance with new Clean Air Act pollution control standards in 2000
  - EPA data :dioxin emissions now account for less than 0.5% of dioxin emissions
- **WTE facilities reduce greenhouse gas emissions. (Climate, global)**
  - EPA estimates :WTE facilities prevent 33 million metric tons of CO<sub>2</sub> per year avoided
- **WTE facilities save real estate. (Land)**
  - They reduce the space required for landfills by about 90%
- **WTE is compatible with recycling. (Resource Minimization)**
  - Communities served by WTE recycle 35% of their trash, compared to 30% for the general population.
  - Annually removes more than 700,000 tons of ferrous materials
  - Nearly 3 million tons of WTE ash is reused as landfill cover, roadbed, or building material.
- **WTE facilities provide economic benefits. (Economic)**
  - WTE is a \$10 billion industry employs ~ 6,000 American workers annual wages ~ \$400 million

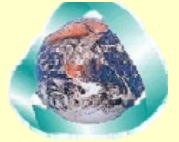


***Boundaries and interfaces of WTE cut across all sustainable fronts***

WTERT 08, New York, NY



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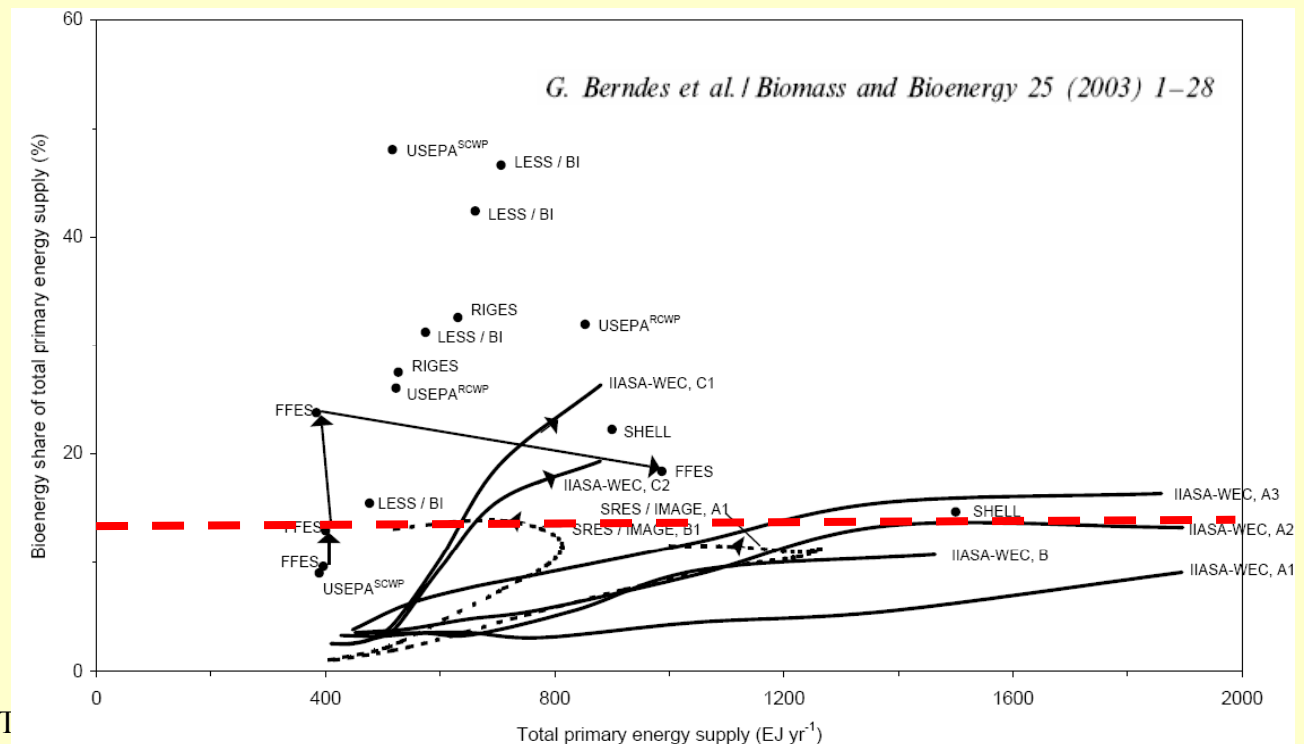
# Bioenergy

- Sustainable carbon neutral power, heat and chemical feedstocks.
- Potential to provide ~ 15% of energy demand.
- Gasification
  - Lower PM, NO<sub>x</sub>, and SO<sub>2</sub>, than combustion
  - Steam addition – Increased concentration of H<sub>2</sub>
  - CO<sub>2</sub> usage – enhanced char conversion and less residue for landfill.
  - Provides reagents (CO & H<sub>2</sub>) for synthetic fuel/chemical production

Estimations vary widely  
due to land availability  
and crop yields.

One piece of the  
alternative energy future  
technologies

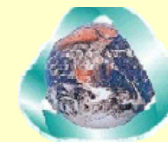
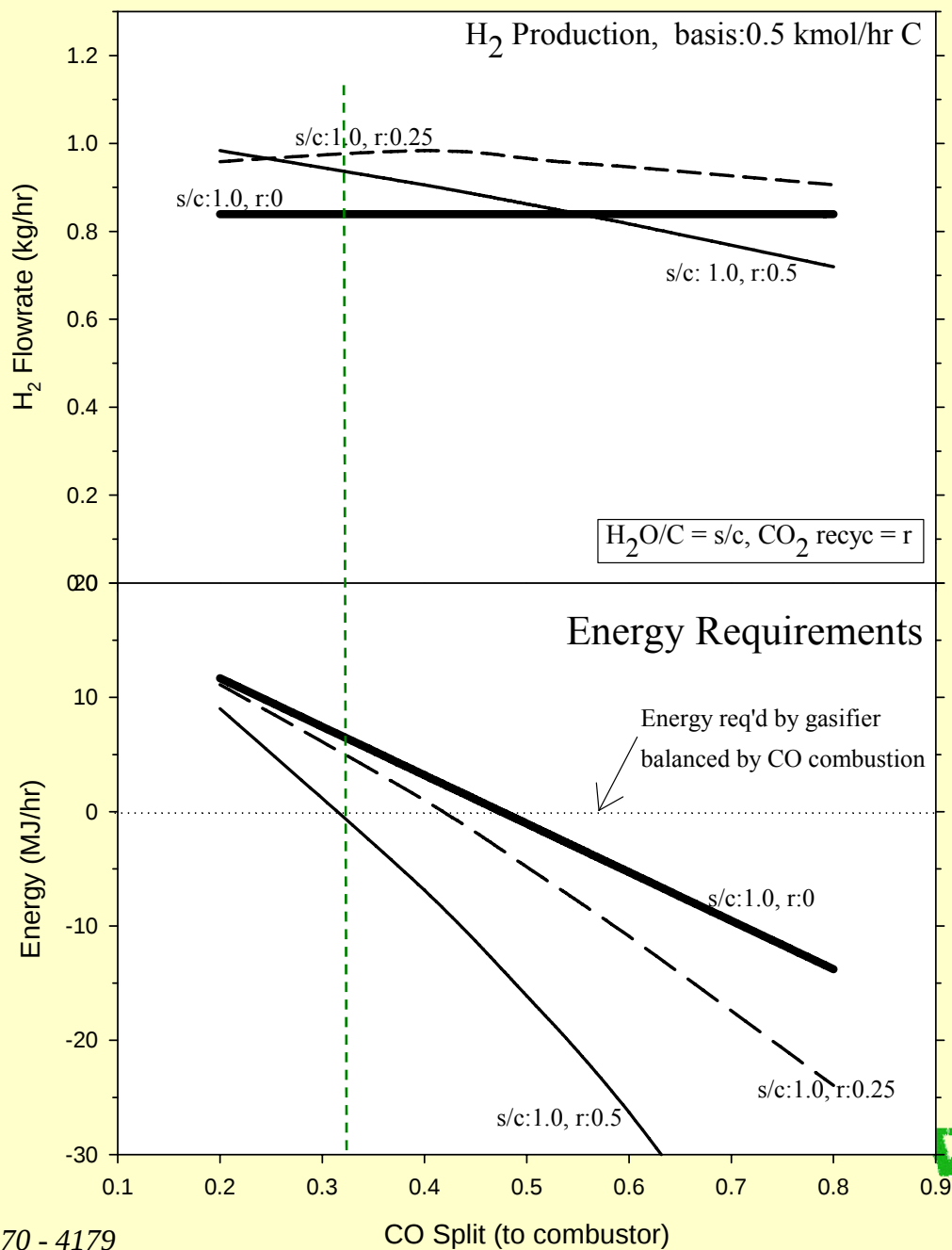
WTERI





***CO<sub>2</sub> recycle comparison for steam to carbon ratio of 1.0.***

**Tradeoff: energy neutral vs H<sub>2</sub> production**



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